

## Synthesis of 1,6-Dioxadispiro[2.0.4.4]dodecan-7-one

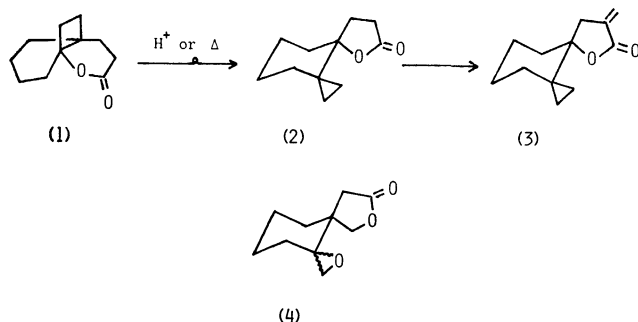
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**Synopsis.** 1,6-Dioxadispiro[2.0.4.4]dodecan-7-one, a useful intermediate of dispiro- $\alpha$ -methylene- $\gamma$ -butyrolactones, was synthesized in a good overall yield by the cycloaddition of dichloroketene to 1,2-dimethylenecyclohexane, and the zinc dust reduction of the chlorine atoms, followed by the Baeyer-Villiger oxidation with *m*-chloroperbenzoic acid.

Recently, much attention has been paid on the chemistry of  $\alpha$ -methylene- $\gamma$ -butyrolactones, especially, concerning the structure-biochemical activity relationship, because a number of naturally occurring sesquiterpene lactones display antitumor and/or cytotoxic activity attributed to this moiety.<sup>1)</sup> From the above point of view, we previously reported on the synthesis of the dispiro- $\gamma$ -butyrolactone (**2**), being a useful intermediate of the dispiro- $\alpha$ -methylene- $\gamma$ -butyrolactone (**3**) having a spiro cyclopropane ring,<sup>2)</sup> by the acid catalyzed or thermally induced cyclobutyl-cyclopropyl-carbinyl type rearrangement of the propellalactone (**1**).<sup>5)</sup> As part of the study on the synthesis of various types of dispiro- $\gamma$ -butyrolactones, we wish to report here a convenient synthesis of 1,6-dioxadispiro[2.0.4.4]dodecan-7-one (**4**) having a spiro epoxide linkage which is expected to enhance the reactivity of the conjugated lactone toward biological nucleophiles.<sup>1a)</sup>

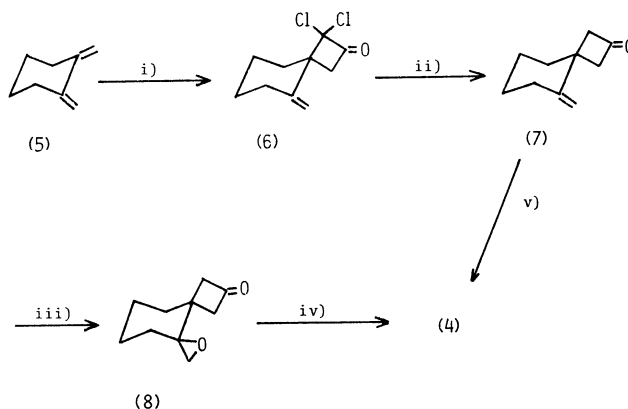


The epoxy- $\gamma$ -lactone **4** was prepared as outlined in Scheme 1; i) the cycloaddition of dichloroketene to 1,2-dimethylenecyclohexane employing the high-dilution method,<sup>6)</sup> ii) the reductive removal of the chlorine atoms with zinc dust-acetic acid,<sup>7)</sup> iii) the epoxidation with *m*-chloroperbenzoic acid (MCPBA), and iv) the Baeyer-Villiger oxidation.

A solution of equal amounts of trichloroacetyl chloride and phosphoryl chloride in ether was added dropwise over a period of 5 h to an ethereal solution of 1,2-dimethylenecyclohexane (**5**), which was readily available from *cis*-1,2-cyclohexanedicarboxylic anhydride, and excess of activated zinc, and the mixture was refluxed for 30 h. The 1:1 cycloadduct, that is the spiro dichlorocyclobutanone (**6**), was obtained in a 61% yield.<sup>8)</sup> The reductive removal of the chlorine atoms from **6** with excess of zinc dust and acetic acid in ether at room temperature gave the spiro cyclo-

butanone (**7**) in a 71% yield. The epoxidation of **7** with 1.2 equivalent of MCPBA in chloroform at room temperature for 3 h afforded the epoxy cyclobutanone (**8**) in a 71% yield.<sup>9)</sup> Finally, the Baeyer-Villiger oxidation of **8** with 1.2 equivalent of MCPBA in chloroform at room temperature for 6 d gave a 1:1 mixture of two epoxy- $\gamma$ -lactones **4**<sup>10)</sup> in a 70% yield. The  $\gamma$ -lactone **4** was also obtained directly from the cyclobutanone **7** by the oxidation with 3-fold excess of MCPBA for 6 d (77%).

In this way, 1,6-dioxadispiro[2.0.4.4]dodecan-7-one **4**, a key compound for the synthesis of dispiro- $\alpha$ -methylene- $\gamma$ -butyrolactones having an epoxide linkage, was synthesized in a good yield by means of the efficient dichloroketene addition.



Scheme 1.

i)  $\text{Cl}_3\text{CCOCl}$ ,  $\text{Zn}(\text{Cu})$ ,  $\text{POCl}_3$ ,  $\text{Et}_2\text{O}$ , 30 h; ii)  $\text{Zn}$ ,  $\text{AcOH}$ ,  $\text{Et}_2\text{O}$ , 30 h; iii) MCPBA (1.2 eq.),  $\text{CHCl}_3$ , 3 h; iv) MCPBA (1.2 eq.),  $\text{CHCl}_3$ , 6 d; v) MCPBA (3 eq.),  $\text{CHCl}_3$ , 6 d.

## Experimental

IR spectra were recorded using a JASCO IR-G spectrometer.  $^1\text{H}$  NMR spectra were obtained on a JEOL JNM-PS-100 spectrometer, using  $\text{Me}_4\text{Si}$  as an internal standard and  $\text{CCl}_4$  as a solvent. MS spectra were measured with a Hitachi RMU-6E spectrometer.

1,2-Dimethylenecyclohexane (**5**)<sup>11)</sup> was prepared in a similar method to that of Davalian *et al.*,<sup>12)</sup> and purified by distillation using a spinning-band distillation column.

1,1-Dichloro-5-methylenespiro[3.5]nonan-2-one (**6**). A solution of 8.1 ml (0.078 mol) of freshly distilled trichloroacetyl chloride and 6.8 ml (0.078 mol) of phosphoryl chloride (distilled from  $\text{K}_2\text{CO}_3$ ) in 440 ml of dry ether was added dropwise over 5 h to a mixture of 8.4 g (0.078 mol) of **5** and 7.6 g (0.114 mol) of activated zinc in 600 ml of dry ether under a nitrogen atmosphere. The reaction mixture was stirred at reflux for additional 30 h. The excess zinc was filtered and washed with ether. The filtrate was concentrated *in vacuo* to ca. 25% of its original volume, an equal volume of pentane added, and

the solution stirred for 1 h to precipitate zinc salts. The solution was decanted from the residue, washed successively with water, a cold saturated  $\text{NaHCO}_3$  solution and brine, and dried ( $\text{Na}_2\text{SO}_4$ ). The solvent was removed *in vacuo* and the residue was distilled under reduced pressure. After recovery of 2.6 g of **5** (bp 60 °C/90 Torr), 7.1 g of **6** was obtained as pale yellow oil (61%): bp 86–88 °C/0.8 Torr; IR 1800, 1635  $\text{cm}^{-1}$ ; NMR  $\delta$  1.16–2.50 (m, 8H), 2.66 (d,  $J=17$  Hz, 1H), 3.62 (d,  $J=17$  Hz, 1H), 4.80 (s, 1H), 5.08 (s, 1H); MS  $m/e$  222 ( $M^++4$ ), 220 ( $M^++2$ ), 218 ( $M^+$ ). Found: C, 54.91; H, 5.56%. Calcd for  $\text{C}_{10}\text{H}_{12}\text{OCl}_2$ : C, 54.82; H, 5.52%.

**5-Methylenespiro[3.5]nonan-2-one (7).** A mixture of 7.0 g (0.032 mol) of **6**, 10 g of zinc, and 10 ml of acetic acid in 150 ml of dry ether was stirred at room temperature for 30 h.<sup>13</sup> The reaction mixture was filtered and the filtrate was washed with saturated  $\text{NaHCO}_3$  solution, brine, and dried ( $\text{Na}_2\text{SO}_4$ ). The solvent was removed *in vacuo* and the residue was distilled under reduced pressure to give 3.4 g of the cyclobutanone **7** (71%): bp 68–69 °C/3 Torr; IR 1775, 1635  $\text{cm}^{-1}$ ; NMR  $\delta$  1.45–1.82 (m, 6H), 2.02–2.28 (m, 2H), 2.45–2.80 (m, 2H), 2.88–3.24 (m, 2H), 4.68 (s, 1H), 4.78 (s, 1H); MS  $m/e$  150 ( $M^+$ ). Found: C, 79.84; H, 9.54%. Calcd for  $\text{C}_{10}\text{H}_{14}\text{O}$ : C, 79.95; H, 9.39%.

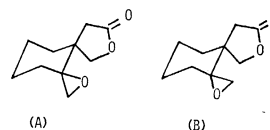
**1-Oxadispiro[2.0.3.4]undecan-6-one (8).** A solution of 250 mg (1.7 mmol) of **7** and 440 mg (2.0 mmol) of 80% MCPBA in 10 ml of chloroform was stirred at room temperature for 3 h. The solution was washed with saturated  $\text{Na}_2\text{SO}_3$  solution, saturated  $\text{NaHCO}_3$  solution, water, and dried ( $\text{Na}_2\text{SO}_4$ ). The solvent was evaporated *in vacuo* to give 197 mg of **8** as colorless oil (71%). Analytical sample was obtained by preparative GLC: IR 1775  $\text{cm}^{-1}$ ; NMR  $\delta$  1.20–2.08 (m, 8H), 2.24–3.12 (m, 6H); MS  $m/e$  166 ( $M^+$ ). Found: C, 72.05; H, 8.70%. Calcd for  $\text{C}_{10}\text{H}_{14}\text{O}_2$ : C, 72.26; H, 8.49%.

**1,6-Dioxadispiro[2.0.4.4]dodecan-7-one (5).** A solution of 100 mg (0.65 mmol) of **8** and 170 mg (0.78 mmol) of 80% MCPBA in 10 ml of chloroform was stirred at room temperature and the progress of the reaction was monitored by GLC. After 6 d, the solution was treated as described above. Chromatography on silica gel (20% ether–petroleum ether) gave 87 mg of **4** as colorless oil (74%).<sup>10</sup> 471 mg of **4** was also obtained in 77% yield from the reaction of 500 mg (3.3 mmol) of **7** and 2.2 g (9.9 mmol) of 80% MCPBA in 27 ml of chloroform for 6 d: IR 1765  $\text{cm}^{-1}$ ; NMR  $\delta$  1.20–2.00 (m, 8H), 2.04–2.72 (m, 4H), 3.60–4.20 (m, 2H); MS  $m/e$  182 ( $M^+$ ). Found: C, 65.72; H, 7.75%. Calcd for  $\text{C}_{10}\text{H}_{14}\text{O}_3$ : C, 65.91;

H, 7.74%.

## References

- 1) a) E. Rodriguez, G. H. N. Towers, and J. C. Mitchell, *Phytochemistry*, **15**, 1573 (1976); b) K. H. Lee, E. S. Huang, C. Piantadosi, J. S. Pagano, and T. A. Geissman, *Cancer Res.*, **31**, 1649 (1971).
- 2)  $\alpha$ -Methylenation of **2** was readily conducted by the usual procedure<sup>3</sup> to afford **3** in a good yield.<sup>4</sup>
- 3) P. A. Grieco, J. A. Noguez, Y. Masaki, K. Hiroi, M. Nishizawa, A. Rosowsky, S. Oppenheim, and H. Lazarus, *J. Med. Chem.*, **20**, 71 (1977).
- 4) Unpublished results.
- 5) a) Y. Tobe, K. Kakiuchi, Y. Kawakami, Y. Sakai, K. Kimura, and Y. Odaira, *Chem. Lett.*, **1978**, 1027; b) K. Kakiuchi, Y. Tobe, and Y. Odaira, *J. Org. Chem.*, **45**, 729 (1980).
- 6) L. R. Krepski and A. Hassner, *J. Org. Chem.*, **43**, 2879 (1978).
- 7) D. A. Bak and W. T. Brady, *J. Org. Chem.*, **44**, 107 (1979).
- 8) The 1:2 cycloadduct was not obtained.
- 9) Trace of **4** was also obtained besides the epoxide **8**.
- 10) Though **4** consisted of two isomers (**A**) and (**B**) they could not be separated completely by GLC or column chromatography; therefore they were characterized as a mixture. Attempt on their separation and structure assignment is in progress.



- 11) P. D. Bartlett, A. S. Wingrove, and R. Owyang, *J. Am. Chem. Soc.*, **90**, 6067 (1968).
- 12) D. Davalian and P. J. Garratt, *J. Am. Chem. Soc.*, **97**, 6883 (1975). **5** was prepared by esterification of *cis*-1,2-cyclohexanedicarboxylic anhydride, reduction with  $\text{LiAlH}_4$ , and treatment with methanesulfonyl chloride followed by reaction with  $\text{KO}^t\text{Bu}$ .
- 13) After 2 h, the monochlorocyclobutanone was obtained as a main product which was purified by preparative GLC: bp 107–108 °C/5 Torr; IR 1785, 1635  $\text{cm}^{-1}$ ; NMR  $\delta$  1.16–2.20 (m, 8H), 2.85 (s, 2H), 4.82 (s, 1H), 4.86 (s, 1H), 4.95 (s, 1H); MS  $m/e$  186 ( $M^++2$ ), 184 ( $M^+$ ). Found: C, 64.90; H, 7.13%. Calcd for  $\text{C}_{10}\text{H}_{13}\text{OCl}$ : C, 65.04; H, 7.10%.